

5-(4-Fluorophenyl)-1-phenylpyrazole

Daniel E. Lynch^{a*} and Ian McClenaghan^b^aSchool of Science and the Environment, Coventry University, Coventry CV1 5FB, England, and ^bKey Organics Ltd, Highfield Industrial Estate, Camelford, Cornwall PL32 9QZ, England

Correspondence e-mail: apx106@coventry.ac.uk

Key indicators

Single-crystal X-ray study

T = 150 K

Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$

R factor = 0.082

wR factor = 0.238

Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The asymmetric unit of the title compound, $\text{C}_{15}\text{H}_{11}\text{FN}_2$, comprises two molecules with dihedral angles of $50.6 (1)/49.2 (1)$ and $51.5 (1)/53.3 (1)^\circ$ between the pyrazole and, respectively, the phenyl and 4-fluorophenyl rings. One intermolecular $\text{C}-\text{H}\cdots\text{F}$ close contact is noted for each unique molecule.

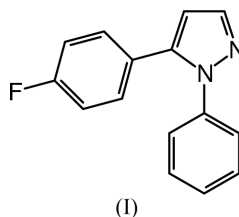
Received 1 May 2002

Accepted 5 June 2002

Online 8 June 2002

Comment

The title compound, (I), was prepared by the addition of phenylhydrazine to 1-(*N,N*-dimethylamino)-2-(4-fluorobenzoyl)ethene. However, in this reaction there are two possible products (Fig. 2). One is formed *via* the hydrazone intermediate which subsequently cyclizes to the 1,3-substituted pyrazole; the other is formed *via* the Michael intermediate, which cyclizes to the 1,5-substituted product, *i.e.* the title compound. The asymmetric unit of this compound comprises two molecules with dihedral angles of $50.6 (1)/49.2 (1)$ and $51.5 (1)/53.3 (1)^\circ$ between the pyrazole and, respectively, the phenyl and 4-fluorophenyl rings. One intermolecular $\text{C}-\text{H}\cdots\text{F}$ close contact is noted for each unique molecule.



Experimental

The title compound was obtained from Key Organics Ltd and crystals were grown from an ethanol solution.

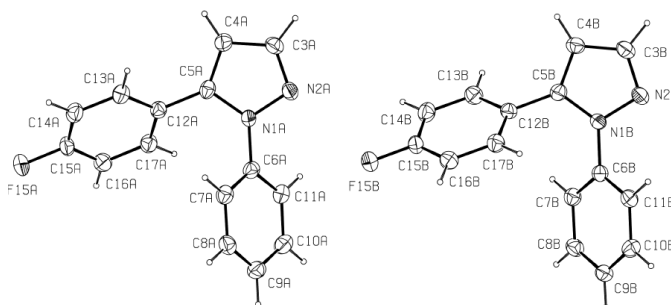
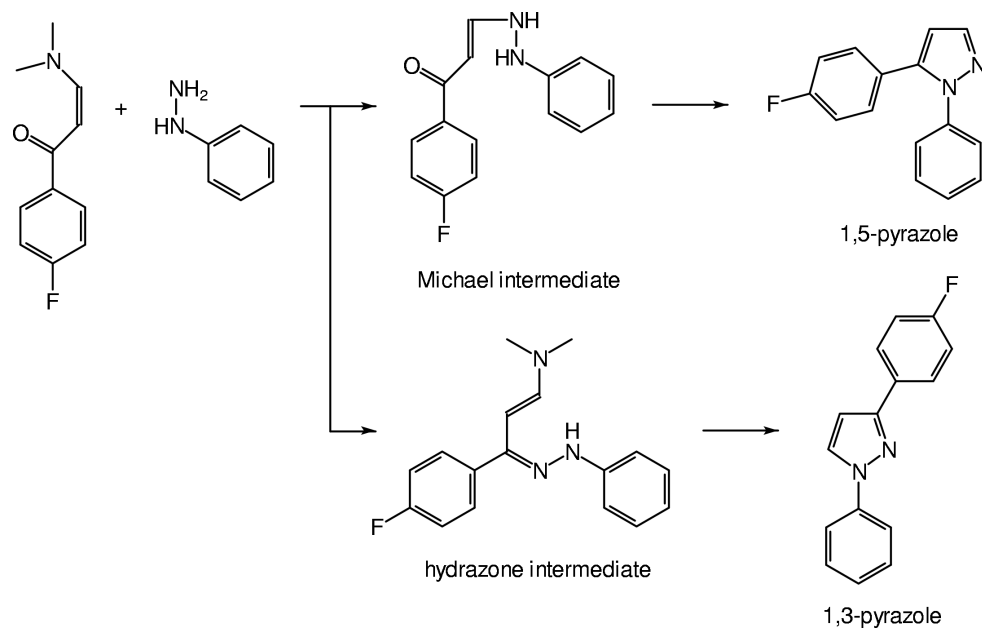


Figure 1

The molecular configuration and atom-numbering scheme for the title compound, showing 50% probability ellipsoids.


Figure 2

Reaction scheme showing two possible intermediates and subsequent products for the reaction of phenylhydrazine with 1-(*N,N*-dimethylamino)-2-(4-fluorobenzoyl)ethene.

Crystal data

$C_{15}H_{11}FN_2$
 $M_r = 238.26$
 Triclinic, $P\bar{1}$
 $a = 7.847$ (2) Å
 $b = 7.904$ (2) Å
 $c = 19.551$ (4) Å
 $\alpha = 90.20$ (3)°
 $\beta = 100.92$ (3)°
 $\gamma = 90.61$ (3)°
 $V = 1190.5$ (4) Å³

$Z = 4$
 $D_x = 1.329$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 16018 reflections
 $\theta = 2.9$ – 27.5 °
 $\mu = 0.09$ mm⁻¹
 $T = 150$ (2) K
 Block, yellow
 $0.20 \times 0.18 \times 0.10$ mm

Data collection

Bruker–Nonius KappaCCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SORTAV; Blessing, 1995)
 $T_{\min} = 0.982$, $T_{\max} = 0.991$
 17989 measured reflections

5311 independent reflections
 3979 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.065$
 $\theta_{\text{max}} = 27.5$ °
 $h = -10 \rightarrow 10$
 $k = -10 \rightarrow 10$
 $l = -23 \rightarrow 25$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.082$
 $wR(F^2) = 0.238$
 $S = 1.17$
 5311 reflections
 326 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0756P)^2 + 2.8942P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.32$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.33$ e Å⁻³
 Extinction correction: SHELXL97
 Extinction coefficient: 0.077 (9)

Table 1

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C3A-H3A \cdots F15A^i$	0.95	2.48	3.415 (4)	169
$C4B-H4B \cdots F15B^{ii}$	0.95	2.44	3.385 (4)	171

Symmetry codes: (i) $-x, -y, 1-z$; (ii) $-x, 2-y, -z$.

All H atoms were included in the refinement, at calculated positions, as riding models, with C–H distances set to 0.95 Å. The R_{int} value of 0.065 was the result of weak high-angle data.

Data collection: DENZO (Otwinowski & Minor, 1997) and COLLECT (Hooft, 1998); cell refinement: DENZO and COLLECT; data reduction: DENZO and COLLECT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON97 (Spek, 1997); software used to prepare material for publication: SHELXL97.

The authors thank the EPSRC National Crystallography Service (Southampton).

References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–37.
 Hooft, R. (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Spek, A. L. (1997). *PLATON97*. Version of May 1997. University of Utrecht, The Netherlands.